

Perpendicular stacking of the triplex and the second helix creates a cleft that functions as the vitamin B12 binding site. Chemical modification studies by Lorsch and Szostak can be interpreted in light of the structure and indicate that the bound and unbound conformations of the aptamer are likely to be fairly similar (most changes in chemical modification upon binding correspond to the protection of residues at the binding interface). Of the 27 nucleotides in the aptamer core (excluding flanking nucleotides required for expression and crystallization), only 4 in helix 2 can be considered to form what would generally be considered conventional secondary structure. The remainder form a variety of interactions that combine to either build the triplex or lock the second helix in position.

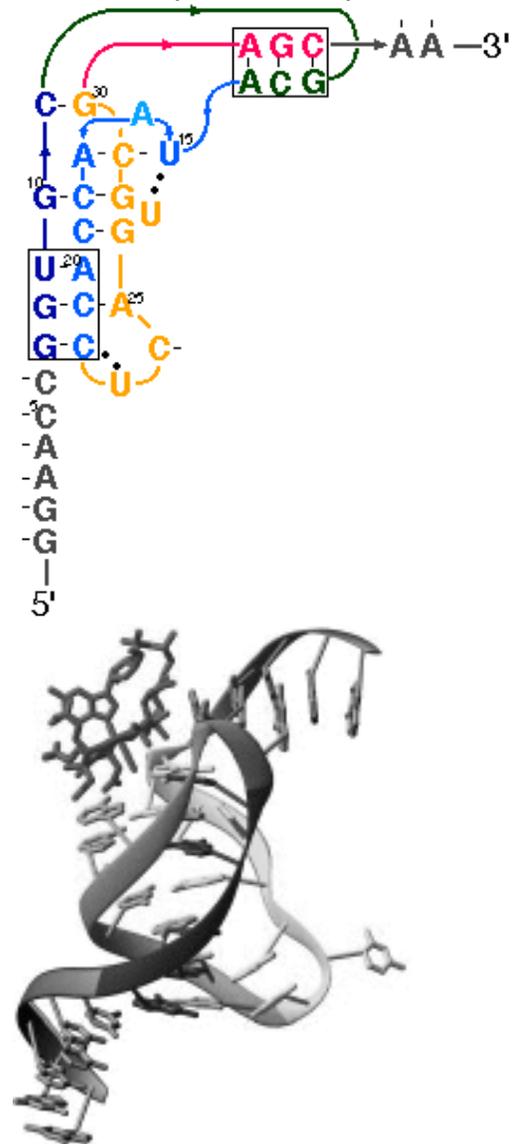
The triplex provides a rich set of tertiary interactions for analysis. Five of the seven tiers in the triplex include base triples stabilized by specific hydrogen bonding. The two triples at the bottom of the triplex (i.e. farthest from the vitamin B12 binding site) are defined by nucleotides packing against the minor groove face of Watson-Crick base pairs. Lying above them in the triplex are two Watson-Crick pairs which mediate a switch in base pairing from strands 1-2 to strands 2-3. As a result of this switch, base triples in the top of the triplex are generally arranged with a nucleotide hydrogen bonding instead to the major groove side of a base pair.

Vitamin B12 binds in the cleft formed at the junction between the triplex and the second helix and makes an extensive set of specific interactions. Methyls protruding axially and equatorially from one side of the corrin ring contact either side of the cleft and make van der Waals contacts with the exposed nucleotide faces of the base triple and base pair that cap the triplex and helix respectively. Supplementing these and additional hydrophobic interactions are a set of four specific hydrogen bonds that bridge the amide side chains of the cofactor and the nucleotides of the RNA.

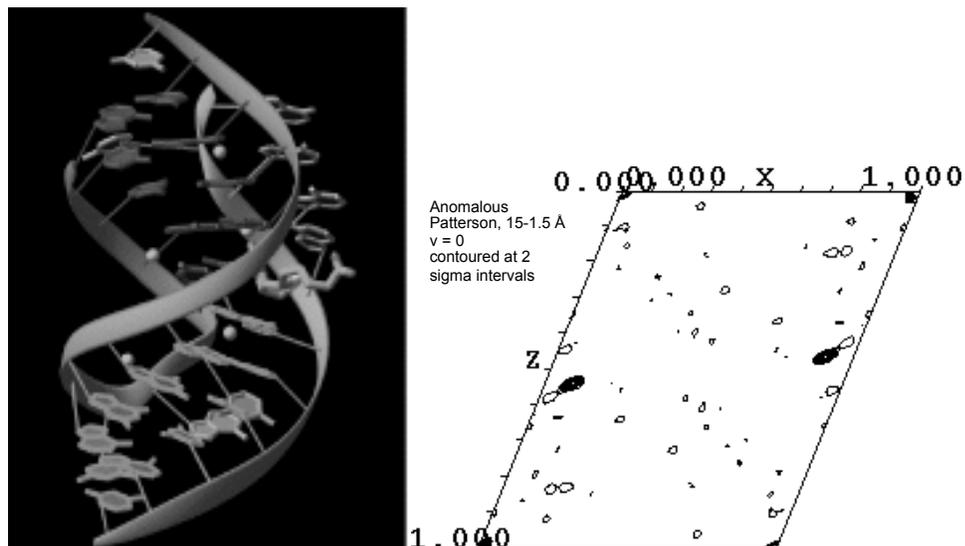
BIOTIN APTAMER

We previously isolated and characterized an RNA motif that specifically recognizes biotin, a ubiquitous carboxylation co-factor used by protein enzymes^{7,8}. A 31-nucleotide pseudoknot, closely resembling the ribosomal frameshifting elements in retroviral mRNAs, is present in all selected RNAs enriched by affinity chromatography. Independent clones containing the pseudoknot bind biotin with relatively high affinity ($K_D \approx 5 \mu\text{M}$) and high specificity. Binding with this avidity is surprising given the absence of charges and aromatic rings on the ligand, functional groups which often provide the driving force for aptamer binding.

Schematic representations of the cyanocobalamin aptamer illustrating its secondary and tertiary structure.



The 1.5 Å crystal structure of the biotin aptamer was determined by MAD phasing using the selenium of bound selenobiotin as an anomalous scatterer. An anomalous Patterson map (inset) clearly shows a single corresponding to the bound cofactor. The refined structure reveals a coaxially stacked pseudoknot, fitting well with previous models constructed on the basis of biochemical and genetic constraints. The arrangement of helices in the aptamer is similar to that observed by NMR and crystallography for viral pseudoknots that induce ribosomal frameshifting. Biotin binds at the interface between the two helices in a pocket defined by the base pairs capping each helix and by the 3'-end of an A-rich loop running the length of the minor groove of helix 1.



REFERENCES

1. A. D. Ellington and J. W. Szostak, "In vitro selection of RNA molecules that bind specific ligands," *Nature* **346** (6287), 818-22 (1990).
2. C. Tuerk and L. Gold, "Systematic evolution of ligands by exponential enrichment: RNA ligands to bacteriophage T4 DNA polymerase," *Science* **249** (4968), 505-10 (1990).
3. N. H. Georgopadakou and A. I. Scott, "On B12 biosynthesis and evolution," *J Theor Biol* **69** (2), 381-4 (1977).
4. S. A. Benner, A. D. Ellington, and A. Tauer, "Modern metabolism as a palimpsest of the RNA world," *Proc Natl Acad Sci U S A* **86** (18), 7054-8 (1989).
5. J. R. Lorsch and J. W. Szostak, "In vitro selection of RNA aptamers specific for cyanocobalamin," *Biochemistry* **33** (4), 973-82 (1994).
6. F.A. Cotton and G. Wilkinson, *Advanced Inorganic Chemistry* (John Wiley and Sons, New York, 1988).
7. Charles Wilson, Jay Nix, and Jack W. Szostak, "Functional requirements for specific ligand recognition by a biotin-binding RNA pseudoknot.," *Biochemistry* **37** (41), 14410-14419 (1998).
8. C. Wilson and J. W. Szostak, "In vitro evolution of a self-alkylating ribozyme," *Nature* **374** (6525), 777-82 (1995).

This work was supported by a grant from the National Institutes of Health (GM52707), the National Science Foundation (MCB-9876350), and an award from the David and Lucile Packard Foundation.

Principle investigator: Charles Wilson, Department of Biology and Center for the Molecular Biology of RNA, University of California at Santa Cruz. Email: wilson@biology.ucsc.edu. Telephone: 831-459-5126.